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# Isotope Notation Worksheet Answers

**examples lithium-7 x 3 li 3 protons 4 neutro s** - any given element can have more than one isotope. to distinguish between the different isotopes of an atom, the element is named with its mass number, for example lithium-7. ... isotope notation name \_\_\_\_ chem worksheet 4-2 7 li 3 examples lithium-7 3 protons 4 neutro s mass number atomic number x a z. **chemistry worksheet name: 235 similar, and how are they ...** - 2. define isotope: 3. the number of protons in an atom is known as the of that atom. 4. the number of determines the name of the atom. 5. the mass number of an atom is the number of plus the number of in the nucleus of the atom. 6. the isotope notation for nitrogen-15 is as follows: a. **isotope notation - california state university, northridge** - isotope notation name \_\_\_\_ chem worksheet 2-1 easy pdf creator is professional software to create pdf. if you wish to remove this line, buy it now. created date: 8/24/2004 12:11:46 pm ... **atoms and isotopes worksheet - tumwater.k12.wa** - atoms and isotopes worksheet 1. fill in the table with the correct information. isotope isotope notation atomic # protons electrons neutrons oxygen-16 bromine-80 uranium-235 copper-64 2. describe the general arrangement of subatomic particles in the atom. 3. **isotope practice worksheet - chemistry** - isotope practice 1. here are three isotopes of an element:  $^{12}_6\text{C}$   $^{13}_6\text{C}$   $^{14}_6\text{C}$  a. the element is: carbon b. the number 6 refers to the atomic number c. the numbers 12, 13, and 14 refer to the mass number d. how many protons and neutrons are in the first isotope? 6 protons & 6 neutrons e. how many protons and neutrons are in the second isotope?

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