Isotope Worksheet And Answer Key

isotope practice worksheet - harrischemistry - isotope name atomic # mass # # of protons # of neutrons # of electrons 92 uranium-235 92 uranium-238 5 boron-10 5 boron-11 . 3. naturally occurring europium (eu) consists of two isotopes was a mass of 151 and 153. ... isotope practice worksheet author: charlene kemmerle created date: 10/11/2011 12:34:18 pm ... *atoms and isotopes worksheet - tumwater.k12.wa* - atoms and isotopes worksheet 1. fill in the table with the correct information. isotope isotope notation atomic # protons electrons neutrons oxygen-16 bromine-80 uranium-235 copper-64 2. describe the general arrangement of subatomic particles in the atom. 3. *isotope practice worksheet - chemistry* - isotope practice 1. here are three isotopes of an element: 12c 14 13c c a. the element is: carbon b. the number 6 refers to the atomic number c. the numbers 12, 13, and 14 refer to the mass number d. how many protons and neutrons are in the first isotope? 6 protons & 6 neutrons e. how many protons and neutrons are in the second isotope? *chemistry worksheet name: 235 similar, and how are they ...* - 2. define isotope: 3. the number of protons in an atom is known as the of that atom. 4. the number of determines the name of the atom. 5. the mass number of an atom is the number of plus the number of in the nucleus of the atom. 6. the isotope notation for nitrogen-15 is as follows: a. *atoms and isotopes worksheet* - atoms and isotopes worksheet 1. fill in the table with the correct information. isotope isotope notation atomic # protons electrons neutrons oxygen-16 bromine-80 uranium-235 copper-64 2. describe the general arrangement of subatomic particles in the atom. 3. *isotopes - practice worksheet - hopkins high school* - isotopes - practice worksheet backg rou nd i nformation : atoms of a given element which have the sdme number of protons but different numbers of neutrons are colled isofopes. thus, isotopes have ihe same position in the periodic table, the some chemicol properties ond the same atomic chorge. *isotope and ions practice worksheet name:* - isotope and ions practice worksheet name: _____ part i: isotopes 1. define an isotope. 2. what would happen if the number of protons were to change in an atom? 3. another way to write isotopes is to write the name of the element then add the mass number after a dash, for example, 14 6 c is carbon-14. *abundance of isotopes name chem worksheet 4-3* - the abundance of each isotope. (remember that the sum of the two abundances must be 100). abundance of isotopes name _____ chem worksheet 4-3 atomic mass = % isotope 1 × mass isotope 1 + % isotope 2 × mass isotope 2 + . . . isotope % abundance mass carbon-12 98.93% 12.000 amu carbon-13 1.07% 13.003 amu *examples lithium-7 x 3 li 3 protons 4 neutro s* - any given element can have more than one isotope. to distinguish between the different isotopes of an atom, the element is named with its mass number, for example lithium-7. remember that the mass number is the number of protons and neutrons. when symbols are used to represent an isotope the mass number is written next to the symbol on the top ... *isotopic abundance practice problems* - ! isotopic abundance - practice problems the atomic mass for each element appearing on the periodic table represents the weighted average of masses for each individual isotope of an element. for example, the atomic mass of carbon is reported as 12.011 amu (atomic mass units). carbon is composed primarily of two isotopes; carbon-12 and carbon-14.

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